

Wen-Jun Zhang,\* Zhong-Yu  
Duan and Xin ZhaoHebei University of Technology, Tianjin  
300130, People's Republic of ChinaCorrespondence e-mail:  
zhang\_wenjun99@163.com

## Key indicators

Single-crystal X-ray study  
 $T = 294$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å  
 $R$  factor = 0.045  
 $wR$  factor = 0.130  
Data-to-parameter ratio = 13.6For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.**(E)-4-[3-Ethoxy-4-(2-phenoxyethoxy)benzylideneamino]-1,5-dimethyl-2-phenyl-1H-pyrazol-3(2H)-one**

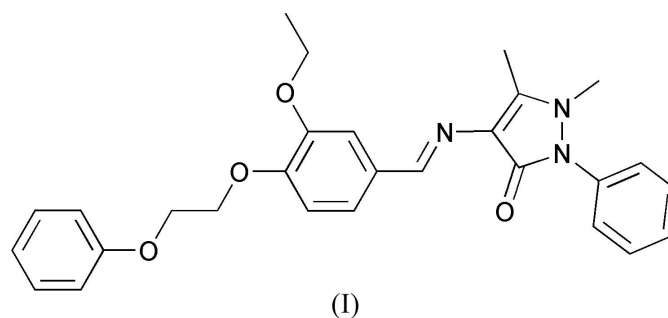
The title compound,  $\text{C}_{28}\text{H}_{29}\text{N}_3\text{O}_4$ , was prepared by the reaction of 4-amino-1,5-dimethyl-2-phenylpyrazol-3-one and 3-ethoxy-4-(2-phenoxyethoxy)benzaldehyde. The vanillin group makes dihedral angles of  $73.08$  (8) and  $66.54$  (6)° with the planes of the two terminal phenyl rings, and an angle of  $30.49$  (6)° with the pyrazolone ring plane. The crystal structure is stabilized by intermolecular  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds that form centrosymmetric dimers.

Received 15 June 2006

Accepted 15 June 2006

## Comment

Metal complexes based on Schiff bases have received a good deal of attention in the development of coordination chemistry because of their significant biological activity (Kahwa *et al.*, 1986). Consequently, a significant effort has been devoted to the synthesis of Schiff base derivatives to develop protein and enzyme mimics (Santos *et al.*, 2001). Among the large number of compounds, 4-amino-1,5-dimethyl-2-phenylpyrazol-3-one forms a variety of Schiff bases with aldehydes, and the synthesis and crystal structures of some them, such as (*E*)-(4)-(3-ethoxy-4-hydroxybenzylideneamino)-1,5-dimethyl-2-phenyl-1*H*-pyrazol-3(2*H*)-one (Han & Zhen, 2005) and (*E*)-4-[4-(4-chlorobenzoyloxy)-3-ethoxybenzylideneamino]-1,5-dimethyl-2-phenyl-1*H*-pyrazol-3(2*H*)-one (Zhang *et al.*, 2006) have been reported. We report here the synthesis and structure of the title compound, (I)



In the title molecule (Fig. 1), bond lengths and angles are within normal ranges (Allen *et al.*, 1987). The pyrazolone ring (C18–C20/N2/N3/O4) is almost planar, with an r.m.s. deviation for fitted atoms of  $0.036$  Å. It makes a dihedral angle of  $46.35$  (7)° with the attached phenyl ring (C23–C28). The vanillin group (C9/C10/C13–C17/O2/O3) is planar, with an r.m.s. deviation for fitted atoms of  $0.014$  Å. This group makes dihedral angles of  $30.49$  (6),  $73.08$  (8) and  $66.54$  (6)°, respectively, with the pyrazolone ring (C18–C20/N2/N3/O4) and the terminal C1–C6 and C23–C28 phenyl rings, respectively.

The crystal structure is stabilized by weak non-classical intermolecular C—H...O=C hydrogen bonds (Table 1) that form centrosymmetric dimers (Fig. 2).

### Experimental

An anhydrous ethanol solution (50 ml) of 3-ethoxy-4-(2-phenoxyethoxy)benzaldehyde (2.86 g, 10 mmol) was added to an anhydrous ethanol solution (50 ml) of 4-amino-1,5-dimethyl-2-phenylpyrazol-3-one (2.03 g, 10 mmol) and the mixture stirred at 350 K for 3 h under N<sub>2</sub>, giving a yellow precipitate. The product was isolated, recrystallized from ethanol, and then dried in a vacuum to give pure compound (I) in 84% yield. Yellow single crystals of (I) suitable for X-ray analysis were obtained by slow evaporation of an ethanol solution.

#### Crystal data

C <sub>28</sub> H <sub>29</sub> N <sub>3</sub> O <sub>4</sub>	V = 1235.8 (13) Å <sup>3</sup>
M <sub>r</sub> = 471.54	Z = 2
Triclinic, P $\bar{1}$	D <sub>x</sub> = 1.267 Mg m <sup>-3</sup>
a = 9.804 (6) Å	Mo K $\alpha$ radiation
b = 10.226 (7) Å	$\mu$ = 0.09 mm <sup>-1</sup>
c = 13.055 (8) Å	T = 294 (2) K
$\alpha$ = 99.985 (11)°	Block, yellow
$\beta$ = 90.178 (11)°	0.30 × 0.24 × 0.18 mm
$\gamma$ = 106.223 (10)°	

#### Data collection

Bruker SMART APEX CCD area-detector diffractometer	6294 measured reflections
$\varphi$ and $\omega$ scans	4325 independent reflections
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)	2786 reflections with I > 2 $\sigma$ (I)
T <sub>min</sub> = 0.961, T <sub>max</sub> = 0.985	R <sub>int</sub> = 0.025
	$\theta_{max}$ = 25.0°

#### Refinement

Refinement on F <sup>2</sup>	w = 1/[ $\sigma^2(F_o^2) + (0.0597P)^2 + 0.2487P$ ]
R[F <sup>2</sup> > 2 $\sigma$ (F <sup>2</sup> )] = 0.045	where P = (F <sub>o</sub> <sup>2</sup> + 2F <sub>c</sub> <sup>2</sup> )/3
wR(F <sup>2</sup> ) = 0.130	( $\Delta\sigma$ ) <sub>max</sub> = 0.001
S = 1.01	$\Delta\rho_{max}$ = 0.16 e Å <sup>-3</sup>
4325 reflections	$\Delta\rho_{min}$ = -0.21 e Å <sup>-3</sup>
319 parameters	
H-atom parameters constrained	

**Table 1**

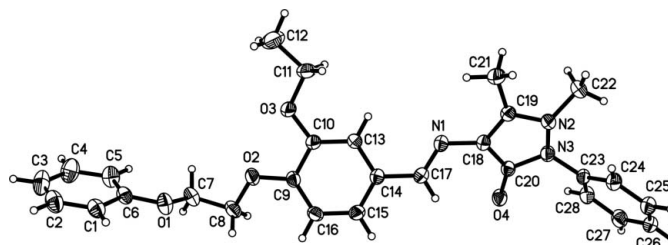
Hydrogen-bond geometry (Å, °).

D—H...A	D—H	H...A	D...A	D—H...A
C17—H17...O4	0.93	2.42	3.049 (3)	125
C27—H27...O4 <sup>i</sup>	0.93	2.56	3.341 (3)	142

Symmetry code: (i) -x + 1, -y - 1, -z.

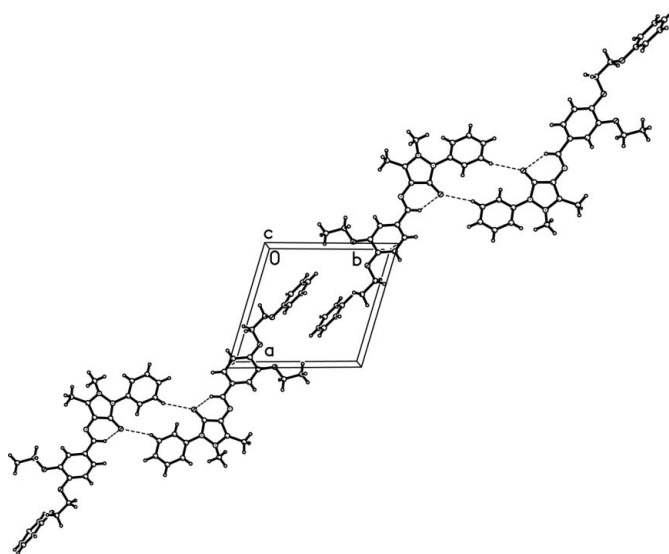
H atoms were included at calculated positions and refined using a riding model. Constrained bond lengths and U<sub>iso</sub>(H) parameters: 0.93 Å and 1.2U<sub>eq</sub>(C) for aromatic, 0.97 Å and 1.2U<sub>eq</sub>(C) for methylene, 0.96 Å and 1.5U<sub>eq</sub>(C) for methyl H atoms. The U<sup>ij</sup> components of atom C12 were restrained to approximate isotropic behaviour.

Data collection: SMART (Bruker, 1999); cell refinement: SAINT (Bruker, 1999); data reduction: SAINT; program(s) used to solve



**Figure 1**

The structure of (I), with displacement ellipsoids for non-H atoms drawn at the 30% probability level.



**Figure 2**

A partial packing diagram for (I), with hydrogen bonds shown as dashed lines.

structure: SHELXS97 (Sheldrick, 1997a); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997a); molecular graphics: SHELXTL (Sheldrick, 1997b); software used to prepare material for publication: SHELXTL.

The project was supported by Hebei Provincial Natural Science Foundation of China (project grant No. 2005000007).

### References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
- Bruker (1999). SMART (Version 5.0) and SAINT (Version 4.0) for Windows NT. Bruker AXS Inc., Madison, Wisconsin, USA.
- Han, J.-R. & Zhen, X.-L. (2005). *Acta Cryst. E* **61**, o3815–o3816.
- Kahwa, I. A., Selbin, J., Hsieh, T. C.-Y. & Laine, R. A. (1986). *Inorg. Chim. Acta*, **118**, 179–185.
- Santos, M. L. P., Bagatin, I. A., Pereira, E. M. & Ferreira, A. M. D. C. (2001). *J. Chem. Soc. Dalton Trans.* pp. 838–844.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997a). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Sheldrick, G. M. (1997b). SHELXTL. Version 5.10 for Windows NT. Bruker AXS Inc., Madison, Wisconsin, USA.
- Zhang, W.-J., Duan, Z.-Y. & Zhao, X. (2006). *Acta Cryst. E* **62**, o2834–o2835.